

## Chapter 12 Assessment Chemistry Answers

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**Stoichiometry: Mass to Mass - Practice - 2** Rust is produced when iron reacts with oxygen.  $4\text{Fe(s)} + 3\text{O}_2\text{(g)} = 2\text{Fe}_2\text{O}_3\text{(s)}$  How many grams of  $\text{Fe}_2\text{O}_3$  are produced when 12.0 ...

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**Calculate Percent Yield with Ideal Stoichiometry - Practice - 1** When 50.0 g of silicon dioxide is heated with an excess of carbon, 32.2 g of silicon carbide is produced.  $\text{SiO}_2\text{(s)} + 3\text{C(s)} \rightarrow \text{SiC(s)} + \dots$

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**Calculate Percent Yield with Ideal Stoichiometry - Practice - 2** If the reaction below proceeds with a 96.8% yield, how many kilograms of  $\text{CaSO}_4$  are formed when 5.24 kg  $\text{SO}_2$  reacts with an ...

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